|   | . Consider the balanced reaction below:  2X (s) + 2H <sub>2</sub> O (l) 2XOH (aq) + H <sub>2</sub> (g)  he species 'X' would be which of the following?  1. a halogen  2. an alkaline earth metal  3. an alkali metal  4. a chalcogen  |
|---|--|
| 2 | <ul> <li>Which of the following is not true of alkaline earth metals?</li> <li>React with halogens to form salts</li> <li>Tend to form a +2 charge</li> <li>Somewhat reactive toward water</li> <li>Gain electrons to achieve noble gas configuration</li> <li>Have 2 electrons in their highest energy shell</li> </ul> |
| 3 | . Which member of the Boron family is a deadly poison 1. Aluminum (Al) 2. Boron (B) 3. Gallium (Ga) 4. Indium (In) 5. Thallium (Tl)  |
|   | . The Nitrogen group contains (1/2) non-metal(s), (1/2) metalloid(s) and (1/2) metal(s). Do not onsider the synthetic superheavy element Ununpentium (Uup) in arriving at your answer.  1. 2, 1, 2 2. 1, 2, 2 3. 2, 2, 1 4. 1, 1, 1 5. 2, 2, 2   |
| 5 | <ul> <li>Which member of the carbon family is most abundant in Earth's crust?</li> <li>1. Silicon (Si)</li> <li>2. Carbon (C)</li> <li>3. Germanium (Ge)</li> <li>4. Tin (Sn)</li> <li>5. Lead (Pb)</li> </ul>   |
| 6 | <ul> <li>Which of the following statements is not true of the oxygen family?</li> <li>1. Are good reducing agents</li> <li>2. They often have an oxidation number of -2</li> <li>3. Contains elements crucial to life</li> <li>4. Are also called chalcogens</li> </ul>  |
| 7 | <ul> <li>Alumina (Al<sub>2</sub>O<sub>3</sub>) is produced in which of the following processes?</li> <li>Hall process</li> <li>Contact process</li> <li>Bayer process</li> <li>Klaus process</li> </ul>  |
| 8 | <ul> <li>Which of the following gemstones is/are derived from aluminum oxides?</li> <li>I. Diamond</li> <li>II. Sapphire</li> <li>III. Ruby</li> <li>1. I</li> <li>2. II</li> <li>3. III</li> <li>4. I and II</li> </ul>   |

5. I and III 6. II and III 7. none